HydroGEN PEC Benchmarking Questionnaire

This survey is designed to collect feedback about best practices for screening and benchmarking of materials, component and devices for PEC water-splitting from the research community. The survey begins with three multiple choice sections that address 1) conditions used during the benchmarking of PEC devices, 2) standard PEC materials, and 3) standard chassis materials/designs. These questions will help develop guidelines for the best practices to use in benchmarking PEC device performance and enable effective comparison of devices across research groups. At the end of the survey, we have also included open questions on critical areas that need further development and how collaboration with national labs can best advance this technology.

After the survey, we would like know whether you would be interested in participating in the development of a Test Framework for PEC water-splitting. If you would like to get involved, we will send out a follow-up email that includes a draft Test Framework and ask for your edits/comments /suggestions. By drawing from the experience of our experts in photoelectrochemistry, we hope to streamline data collection and foster a collaborative environment that leads to new breakthroughs in PEC water-splitting.

* Required

1. Email address *

2. Please list your name: *

3. Please list your affiliation *

What standard conditions should we use to benchmark devices for unassisted photoelectrochemical water splitting?

Background and motivation: We aim to develop standards for benchmarking performance, so comparisons between devices from different research groups can be made in the future. In addition to device-specific optimal operating conditions, community-accepted standard tests, developed through this exercise, are strongly encouraged to include in publications.

4. 1.) Do you think reporting the performance of devices at standard conditions, in addition to "favored" testing conditions, would be useful?

\bigcirc	Yes
\bigcirc	No

	5.	5. If no, please explain:	
 water splitting? If so, what MINIMUM photoelectrode area should ALWAYS be reported for benchmarking? (choose one) Mark only one oval. A minimum size is unnecessary 0.1 cm2 0.5 cm2 1.0 cm2 Option 5 Other: 7. 3.) Should we choose one or several standards for operating pHs? If so, what pH values should ALWAYS be reported for benchmarking? (choose all that apply) <i>Check all that apply.</i> 0			
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should ALWAYS be reported for benchmarking? (choose all that apply) Check all that apply. 0		Other:	
	7.	should ALWAYS be reported for benchmarking? (cr	
		0	

0
7
14
Other:

8. If no, please explain:



9. 4.) What electrolyte(s) should ALWAYS be reported for benchmarking in acidic conditions? (choose all that apply)

Check all that apply.

H2SO4
HCIO4
HCI
Other:

10.	5.) What electrolyte(s) should ALWAYS be reported for benchmarking in neutra	al
	conditions? (choose all that apply)	

Check all that apply.

Phosphate buffer solution	
Borate buffer solution	
Un-buffered KCI or NaCI solution	
Other:	

11. 6.) What electrolyte(s) should ALWAYS be reported for benchmarking in basic conditions? (choose all that apply)

Check all that apply.

NaOH			
КОН			
Other:			

12. 7.) Should standard illumination conditions be reported for benchmarking? If yes, what illumination intensities should ALWAYS be reported for benchmarking? (choose all that apply)

Check all that apply.

A standard illumination is necessary
0.1 Sun
1 Sun
10 Sun
Other:

13. 8.) What MINIMUM number of diurnal cycles should ALWAYS be reported for benchmarking? (choose one)

\bigcirc	Diurnal cycling is unnecessary (please explain below)
\bigcirc	1
\bigcirc	2
\bigcirc	4
\bigcirc	20
\bigcirc	40
\bigcirc	Other:

ALWAYS be r	eported for be	ark operating enchmarking	temperatur ? (choose a	e? If so, w II that app	vhat temp bly)	eratures should
Check all that	apply.					
~30 C (ro	oom temperatu	ıre)				
~70 C (e	levated temper	rature)				
Other:						
			k operating			
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for ALL benc Check all that None, S ⁻ Total hyc Faradaic Spectral Other:	hmarking con apply. TH is sufficient rogen produce Efficiency for responses	ditions? (cho ed in kg HER and OEF	to-hydrogen	apply)	-	ALWAYS be repo

Skip to question 18.

What standard materials would be the most useful?

Background and motivation: Working with HydroGEN Lab nodes, we aim to develop standard

materials and/or devices that can be used to compare conditions between different labs and enable rapid prototyping.

19. 1.) Would standard light absorbers that produce enough voltage for unassisted water splitting be useful for testing catalysts or protecting layers? If yes, which would be most useful?

Mark only one oval.

Triple junction amorphous silicon? (e.g., V(OC)=2.2 V, J(SC)=7 mA/cm2, FF=0.57) (J.Jin et al., Energy Environ. Sci., 2014, 7, 3371-3380 and SY Reece et al., Science, 2011, 334, 645-648)

Tandem junction III-V (e.g., V(OC)=2.4 V, J(SC)=7.6 mA/cm2, FF=0.76) (E. Verlage et al., Energy Environ. Sci., 2015, 8, 3166-3172)

Triple junction III-V from Spectrolab Inc.? (e.g., V(OC)=2.55 V, J(SC)=14.85 mA/cm2, FF=0.7) (K. Walczak et al., Adv. Energy Mater., 2017, 7, 1602791)

Custom tandem III-V (V(OC)>2.0 V, J(SC)=11.5 mA/cm2) (JL Young et al., Nature Energy, 2017, 2, 17028)

- Other:
- 20. 2.) Would a standardized photocathode or photoanode be useful to characterize or integrate catalysts or protective coatings? If yes, which would be most useful? (J-V characteristicss for the referenced electrodes are shown below)

Mark only one oval.

n-p+-Si photoanode (V(OC)=0.55 V, J(SC)=33.6 mA/cm2, FF=0.29) (see S.Hu et al., Science, 2014, 344, 1005-1009)

p-n+-Si (n(D)=not reported, V(OC)=0.58 V, J(SC)=30 mA/cm2, FF=0.58) (see M. Kast et al., ACS Appl. Mater. & Inter., 2014, 6, 22830-22837)

p-Si (V(OC)=0.37 V, J(SC)=22.7 mA/cm2, FF=0.58) (see E.L. Warren et al., J. Phys. Chem. C, 2011, 115, 594-598)

n-Si (n(D)=1x10(19) cm(-3), V(OC)=0.55 V, J(SC)=34.7 mA/cm2, FF=0.29) (see S.Hu et al., Science, 2014, 344, 1005-1009)

np+-GaAs (n(D)=5x10(17) cm(-3), V(OC)=0.77 V, J(SC)=33.6 mA/cm2, FF=0.86) (see S. Hu et al., Science, 2014, 344, 1005-1009)

p-GaAs n(D)=1x10(17) cm(-3), V(OC)=0.7 V, J(SC)=22 mA/cm2, FF=not reported) (see JL Young et al., J. Mater. Chem. A, 2016, 4, 2831-2836)

n-GaAs (n(D)=5.5x10(16) cm(-3), V(OC)=0.7 V, J(SC)=20 mA/cm2, FF=0.6) (see F.Yang et al., J. Phys. Chem. C., 2016, 120, 6989-6995)

Not Useful

Other:

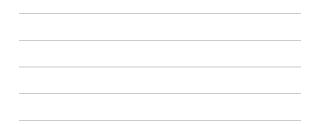
21. 3.) Would you find any of these catalysts useful as a standard dark anode for OER to photocathodes?				
	Mark only one oval.			
	None, these are readily available.			
	Ni/NOx for pH=14			
	IrO2 for pH=0			
	RuO2			
	NiFeOx			
	Other [.]			

22. 4.) Would you find any of these catalysts useful as a standard dark cathode for HER to test photoanodes?

Mark only one oval.

\bigcirc	None, these are readily available
\bigcirc	Pt
\bigcirc	Pd
\bigcirc	Ni/NiOx
\bigcirc	Other:

23. 5.) Comments and/or questions that we missed in this topic?



What sort of standard chassis would be the most useful?

Background and motivations: We aim to design a standard chassis that will facilitate rapid testing of devices. The goal would be to widely distribute these and ensure benchmarking is consistent as possible. Depending on the cost of production, we may be able to distribute these beyond the labs directly involved in this initiative.

24. 1.) Would a standardized chassis design be useful?

Yes		
No		
Other:		

25	lf	no	please	exp	lain [.]
Z J.		пυ,	please	evh	ann.

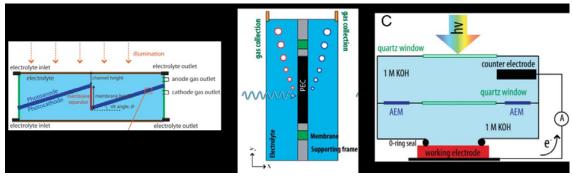
2.) What chassis material should be used for I one) Mark only one oval.	penchmarking in acidic solutions? (choose
Teflon	
Acrylic	
Polycarbonate	
High density polyethylene	
Other:	

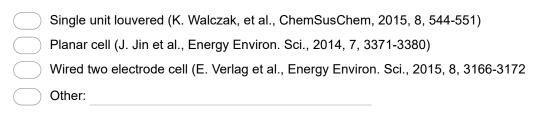
27. 3.) What chassis material should be used for benchmarking in basic solutions? (choose one)

Mark only one oval.

\bigcirc	Teflon
\bigcirc	Acrylic
\bigcirc	Polycarbonate
\bigcirc	High density polyethylene
\bigcirc	Other:

28. 4.) What chassis style would your lab like to work with?





29. 5.) What maximum price range would you be willing to pay for a standard cell? Labs participating in HydroGEN should be provided cells as part of the initiative, but we would like to gauge whether these can be produced at a price that other labs could afford. (choose one)

Mark only one oval.

Less than \$10
\$10-\$30
\$30-\$50
\$50-\$100
More than \$100
Price is not the first consideration for us
Other:

- 30. 6.) What is the price of the current cell your lab uses to test photoelectrochemical devices? It would be useful to consider designing cells that would be a competitively priced alternative that labs outside the initiative would purchase.
- 31. 7.) Other things that you would like to know, please list.

OPEN QUESTIONS

- 32. **1.) What are the most pressing needs/challenges for PEC water splitting?** *Mark only one oval.*
 - Lack of suitable abundant materials
 - Device stability
 - STH efficiency
 - Cost per kg of H2
 - Other:

33. 2.) What are the critical parameters to calculate and characterize for PEC?

34. 3.) How can we accelerate testing of device stability?

35. 4.) What additional techniques/instruments/capabilities would be most useful for the HydroGEN consortium to develop (see existing capabilities at https://www.h2awsm.org/)?

36. 5.) What is the most immediate way to address the scale up challenge (elaborate a bit on the gaps)?

37. 6.) Additional questions or comments regarding PEC water-splitting?

Feedback on Test Framework

38. Would you like to get involved in developing the Test Framework for PEC water-splitting? and here is what the Test Framework might look like. *

Material properties

Class of Material	Key Parameters	Standard	Techniques	References	Notes/Limitations
P	Bandgap	Si: 1.1eV ¹	UV-vis	2.3	-Subjective analysis -Bulk band gap may differ from surface
			PL	4	-Shows optimum performance
	Band positions (valence band/conduction band)	n-Si (111): $E_{vbm}^B = 0.88 \text{ eV}^3$ p-Si (111): $E_{vbm}^B = 0.27 \text{ eV}^3$	EIS (Flat band potential)	2	-Surface heterogeneity -Impacted by surface states -No spatial resolution
			XPS/UPS	6,7	-Ex situ
Photo-absorber	Minority carrier diffusion length (carrier mobility, carrier life time)	$\begin{array}{c} p\text{-Si} \; (B\text{-doped}, n_D = 6 \times 10^{15} \; cm^{-3}); \\ 250 \; \mu m \\ n\text{-Si} \; (P\text{-doped}, n_D = 5 \times 10^{16} \; cm^{-3}); \\ 168 \; \mu m^{-8} \end{array}$	Transient absorption spectroscopy	9	-Measures lifetime only
			Time-resolved Photoluminescence	10	-Cryogenic temperatures -Non-radiative sources of decay may contribute
			Electron Beam- Induced Current	п	-Typically qualitative -Damages organic materials -Ex situ
			Chopped photocurrent-time	12	-Low precision
		p-Si: B, 6×10 ¹⁵ cm ⁻³ n-Si: P, 5×10 ¹⁶ cm ^{-3 ±}	EIS (Mott-Schottky)	2,9 13	-Amorphous structures -Surface states
	Doping types and doping concentrations		Hall measurements		-Low mobility -High polycrystallinity -Multiple carrier species
	Photo-generated carrier		EQE	23,14,15	-Sensitive to lamp calibration
	collection efficiency		IQE		-Determination of absorbed photons difficult

Mark only one oval.



A copy of your responses will be emailed to the address you provided

