

Advanced Water-Splitting Technology Pathways Benchmarking & Protocols Workshop

Breakout Session Summaries Low Temperature Electrolysis (LTE)

March 2 – 3, 2021

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Session ID	Topic	Lead
Session ib	Торіс	Leau
LTE-1	LTE Technology Roadmap Review & Discussion- Catalysts	Shannon Boettcher (Univ of Oregon)
LTE-2	Technology Roadmap Review & Discussion- Porous Transport Layer (PTL) Tech	Nemanja Danilovic (LBNL)
LTE-3	Techno-Economic Analysis - LTE	Brian James (Strategic Analysis, Inc)
LTE-5	LTE Cell Test Methods & Reference Cell	Marcelo Carmo (Juelich)
LTE-7	Technology Roadmap Review & Discussion - Membranes	Andrew Motz (Nel Hydrogen)



Session ID: LTE-1

Title: LTE discussion - Catalysts

Summary of discussion

- Catalysts for PEM and AEM
- 2. Ionomer needs in catalyst layer
- 3. How to distinguish catalyst and ionomer contributions to performance in catalyst/ionomer layer
- 4. Understanding H⁺/OH⁻ concentration control within catalyst/ionomer layer

Consensus and/or dissenting opinions

- Consensus Ir and Pt catalysts work for PEM. Direction should be toward lower loadings rather than non-PGM in acid
- 2. Dissenting opinions should AEM be compared to PEM or liquid alkaline?

Key Take-Aways

- Better catalyst dispersion is key to achieving lower loadings
- 2. The durability of Ir in PEM systems with lower loadings (< 1 mg cm⁻²) needs to be understood in MEA configuration
- 3. Need to close gap between AEM performance and durability



Session ID: LTE-1

Title: LTE discussion - Catalysts

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Session ID: LTE-2

Title: Porous Transport Layer (PTL) Tech Roadmap

Summary

- Commercially available materials include Bekaert (fiber), Mott (sinter) serve as baseline
 - Performance differences and optimal structures have not been determined. This is what we have based on filtration industry
 - -We still don' tknow what the optimal structure might be wrt thickness, porosity, particle size etc
 - -The interface between PTL/CL must be thoroughly interrogated
- · Materials needed to advance PTLs:
 - Need to have similar performance and durability, mechanical properties, Changes away from Ti PTL materials need to consideration durability, cost to upscale and coating lifetime if applicable
 - Need to understand how low in PGM coating and uniformity can go or replace with PGM-free if equivalent performance and durability
 - -For coatings how long does it take to apply and what is the cost of the process in addition to PGM content cost
- The mechanical properties of PTLs are important to withstand compressive forces acting on it and its land/channel support. Need to define the mechanical failure mode and then testing to find the metric to assess.
- The corrosion testing in aqueous environments is more aggressive than in the MEA based on local environment measurements (pH and V)
- MPLs are common in fuel cells but have not been optimized for electrolyzers, nor is there consensus on their ultimate need.
 - -MPLs may prove useful for stability of PTLs if thinner PTLs are used
 - -They may hinder oxygen removal and water flow too much.

Consensus/Dissenting

- Academia desires more specific design parameters from industry to lead investigation into new/different materials and requirements of the materials in service conditions with proprietary cells
- Researchers could/should create benchmark tests that would measure fundamental properties that are of interest to industry.
- -Mechanical properties
- -Corrosion assessment of coatings
- -Coating uniformity requirements and specs
- More information is still needed in regard to the usefulness of MPLs for electrolysis.
 - -Wrt to two phase flow and catalyst layer electronic conductivity, as well as swelling of the membrane/ionomer into PTL

Key Take-aways

- When investigating new materials, the function of the material, durability and cost at scale-up need to be considered.
- We don't know what the ideal PTL structure is that we would want to scale in production (beyond current baseline)
- •The mechanical properties of PTLs is just as important as it's other functions as a cell component.
- •The usefulness of MPLs is still questionable.
- Differential pressure electrolyzer and un-pressurized system will have different PTL needs

- 1. Determine mechanical property test for PTL
- 2. Determine corrosion resistance test for PTL and coating
- 3. Determine two phase flow properties of PTL
- 4. Determine properties and limitations of PTL/CL interface
- 5. Determine PTL structure function relationships
- 6. Determine MPL sturture/function relationship
- 7. Find alternative PTL materials
- 8. Find alternative PGM-free coating materials, or minimum required PGM



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Title: Porous Transport Layer (PTL) Tech Roadmap

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Session ID: LTE-2

Title: Porous Transport Layer (PTL) Tech Roadmap

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Session ID: LTE-3

Title: LTE TEA Breakout

Summary of discussion

- Questions of current analysis
 - Are fixed oper. costs correct? (PEM should be lower than ALK)
 - Discussion of pressure, thinner membranes, and X-over
- Desire to lowering PGM usage
- Plate coatings: current options
- AEM: durability vs. capex tradeoff
 - What is voltage goal at 1A/cm2 (with Pt vs. non-PGM cathode)

Key Take-Aways

- Need thinner membrane (but must watch X-over and affect on polarization curve)
- Pt is costly, but most promising to reduce rather than replace with non-PGM
 - Reduce as coating and catalyst
- Ir needs to be reduced 10x
- Decent base case but many unanswered trade-offs:
 - Durability vs. PGM loading
 - Current Density vs. CapEx vs. Efficiency
 - Cheap Electricity vs. Capacity Factor

Consensus and/or dissenting opinions

- For base material that will corrode, coatings need to be perfect
- Getting beyond Ir and Ti will be difficult for PEMthrifting or supporting such materials may allow cost reduction
- May not be able to eliminate PGM. But how low is low-enough?
- Maintain PGM coatings performance while reducing loadings (and thus costs)
- 2V in oxygen rich environment is a big challenge

- Explore manufacturing development to drive down costs. (3D printing, molding, etc.)
- Try to learn-from/make-use-of solar industry lessons
- Need to understand durability:
 - impact of alternate/cheaper materials,
- Need to stabilize OER catalyst during on/off cycling (which will be more sensitive to catalyst dissolution at low loadings)



Session ID: _LTE-3

Title: LTE TEA Breakout

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Session ID: LTE-5

Title: LTE Cell Test Methods & Reference Cell

- Summary of Discussion
- MEA and CCMs
- How should MEAs/CCM be stored?
- Storage depends on the processing and the size.
- Humidity should also be considered as the membranes can/will expand with hydration and if the hydration needs to be controlled this could be a concern.
- There are some industry protocols that exist for "cleaning" MEAs after longer term storage.
- Can or should components be reused?
- Typically this is only done out of necessity.
- EIS measurements should always be taken if reusing components to evaluate the appropriateness of this process.
- Test cells need to work for both performance and degradation experiments, which might warrant research into different cell materials.
- Is there any need to further refine harmonized test protocols for pol. Curves?
- Possibly reverse steps 1 and 2 of the protocol or add a step prior to step 1.
- Step 2 should be reworded as either, optional, relevant or based on manufacturer requirements. Any processing of the membrane done by researchers, such as applying electrodes, could also affect this step.
- Better definition of the acceptable deviation in water flow, temperature and pressure is needed.
- A better definition of when operating conditions reach stability is needed.
- Temperature should be reported as inlet and outlet temperature, rather than just temperature in general.



Session ID: LTE-5

Title: LTE Cell Test Methods & Reference Cell

Key Take-Aways

- The importance of harmonizing:
 - Test protocols
 - Reference components and cells
 - Performance and durability data
- Create trust and allow meaningful comparability of results (apples with apples)
- Aim is to create a simple reference point for data comparison within and outside a given institution
- Current activities (interest) within:
 - DOE (2B, H2New), IEA-Annex30, EU-JRC, DOE
- Importance to keep the continuity of such initiatives and its harmonization

Consensus (C) and/or Dissenting (D) Opinions

- C- To adopt the earlier results/experiences within Annex30
- **C-** Further refining of defined protocols important
- D- Not clear how valid the information is, since technology is under constant development
- C- Agreed that such protocols need to be re-visited after a given period
- C- Join forces (H2New + Annex30) to allow continuity and makes use of synergy

- Continue discussion in a smaller group (clear reporting information, break-in procedure, flow, EIS)
- Intensify discussion to define the strategy to combine efforts (H2New + Annex 30, and others)
- Definition of common protocol and a reference cell for PEM electrolyzers
- Work on possible MEA alternatives as reference (challenges with ordering E300 from Greenerity)
- Long-term testing and eventually ASTs
- Brainstorm the possibility to start a harmonization of AEM electrolyzers
- List of key names and contact points



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Session ID: LTE-5

Title: LTE Cell Test Methods & Reference Cell

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PEMWE

Session ID: LTE-7

Title: Technology Roadmap Review &

Discussion- Membranes

Summary of discussion

Durability

- o Compression and tension in polymers are different processes and have different failure modes
- During testing need to take into account: active area is under different pressure than the area directly under the seal; pressure fluctuates during testing (can lead to membrane puncturing)
- Fluoride leaching not detected when measured at the anode, so the chemical degradation of the membrane is not confirmed. However, leached cations will likely cause damage to the membrane
- Initiative to use thinner membranes to improve cell resistance (thinner membrane = less resistive) can result in lack of durability due to poor mechanical properties. Potential solution: to use reinforced membranes. But: mechanical reinforcement can decrease conductivity
- $\circ~$ Going below 50 μm in membrane thickness might improve performance but decrease durability Characterization
- Tests done ex situ on dry membranes are not relevant. Ex situ tests on hydrated membranes under elevated temperatures are complicated
- H2 crossover rate is assumed to be constant at constant pressure, but some data reports suggest a dependence on current
- o H2 sensors get wet during operation and cannot measure H2 levels
- o Thinner membranes will result in more H2 crossover

Key Take-Aways

- To achieve performance targets, we should go as thin as possible – we need to investigate performance and durability trades of thinner membranes
 - Mechanical reinforcement trade-offs should be considered
 - $\circ~~50~\mu m$ might be an optimal thickness for the membrane
- H2 permeation will become increasingly important to understand as thinner membranes are used
 - High H2 crossover leads to cell efficiency loss that has to be offset by operating at high current density
- Lack of ex situ and in situ tests for the mechanical properties of membranes (specifically compression tests)

Consensus and/or dissenting opinions

- Does the observed amount of H2 crossover presents a safety concern?
 - ✓ Yes it might for thinner membranes
 - ✓ No, the bigger concern is cell efficiency loss
- Should we aim for thinner membranes or stick to the current optimal thickness?
 - ✓ If performance suffers from reinforcement, no point in using thinner membranes
 - ✓ Reinforcement might improve mechanical properties of thinner membranes (same performance as thicker membranes but better durability)

- Need to develop thinner, robust, electrolysis specific membranes
- Develop ex-situ testing that is representative of common failure modes
- Understand the phenomena that is causing observed increase in H2 permeation



AEMWE

Session ID: LTE-7

Title: Technology Roadmap Review &

Discussion- Membranes

Summary of discussion

Supporting electrolytes

- o improve performance of non-PGM catalysts
- Use of highly concentrated KOH requires hazmat precautions and regular maintenance; leads to membrane degradation through OH- attack or precipitation
- K2CO3 is less corrosive than KOH. But: causes pH gradients in the membrane, CO2 release, and precipitation. pH at the cathode is high even in a dilute electrolyte
- For the purpose of advancing technology might stick to using KOH if cheap H2 can be produced
- Supporting electrolyte gets access to more active site: better catalyst utilization
- Another strategy to improve non-PGM AEMWE performance: new ionomer/membranes and GDL design
- Increasing the stack number to achieve the performance of PEMWE may not be cost-effective even though each cell stack alone is cheaper in materials than a single PEM stack

Consensus and/or dissenting opinions

- Are shunt currents an issue for the system?
 - ✓ No, system design allows to avoid them at a low cost

Key Take-Aways

- Separation of pure-water systems and systems that use supporting electrolyte: different design principles and considerations
 - o Systems with supporting electrolyte are more of a near term technology
- For AEMWE with supporting electrolytes, it is preferred to use dilute electrolyte to avoid related issues
- Some membrane degradation issues might be solved by testing cathode feed
- The investigation into pH gradient in the system is desirable
- There is no AEM equivalent to Nafion in performance and durability

Action Items

 Perform a TEA analysis on Stack cost, durability, and current density (assume voltage parity with current PEM ~1.9 V) to understand the minimum requirements to achieve cost parity.



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Title: Technology Roadmap Review & Discussion- Membranes

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