



# HydroGEN

Advanced Water Splitting Materials

## Successful Examples of Project – HydroGEN STCH Nodes Interactions

Presented by Anthony McDaniel

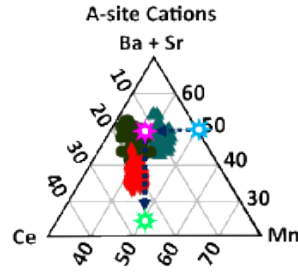
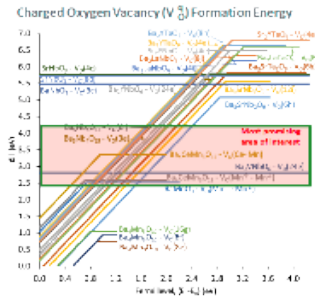
DOE Hydrogen Program Kick-Off Meeting, August 21-22, 2023

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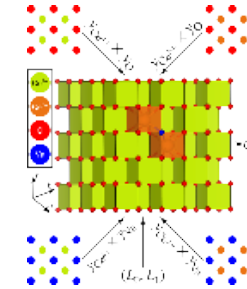
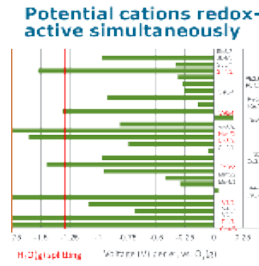


# STCH Seedling Projects are Fulfilling the Vision of the Consortium/EMN Model (HPC, ML, theory guided material design)

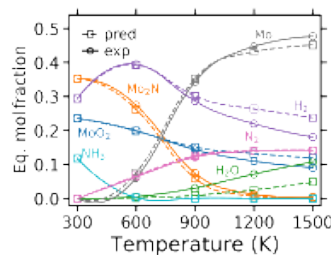
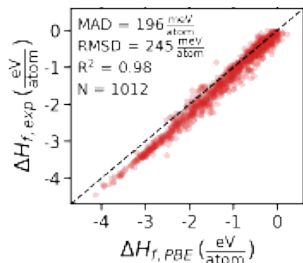
- Found RP phases that modify redox thermo.
  - DFT screening of defect formation energy
  - Thin film combinatorics for compound discovery
  - High throughput colorimetric screening



- Incorporate second redox active sublattice to modify thermo.
  - DFT method to predict  $\Delta\delta$  a priori using simple sublattice model formulations
  - Discover compounds with optimized thermo ( $\delta H$ ,  $\delta S$ )

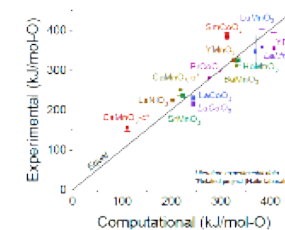


- Use machine-learned models coupled to DFT to discover new redox materials.
  - Rapidly screen materials based on machine-learned predicted stability
  - Formulate descriptor(s) for predicting reaction network energetics and equilibrium



Northwestern University

- Use high-throughput Density Functional Theory to discover new redox materials.
  - Screen  $>10^4$  known compounds for ground state stability/synthesizability and favorable thermo at reduction  $T < 1400$  °C



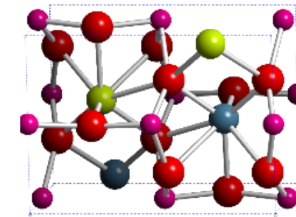
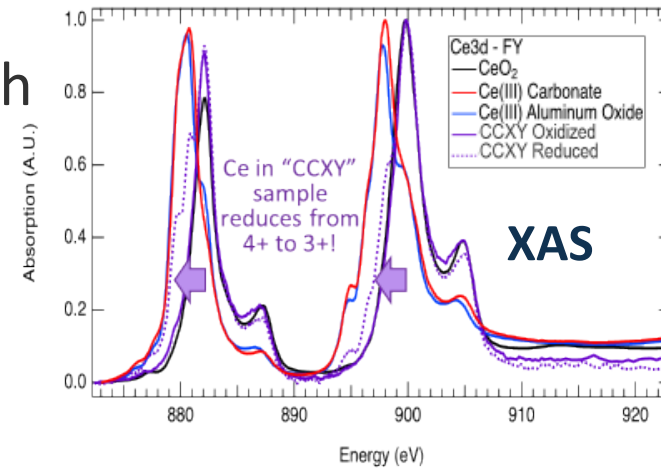
- One dozen *potential* STCH compounds have been “discovered” using HPC, ML, and DFT
- Water splitting functionality has been verified in several of these predicted formulations
- Validated high-throughput computational tools are now in place to rapidly expand the known STCH material space



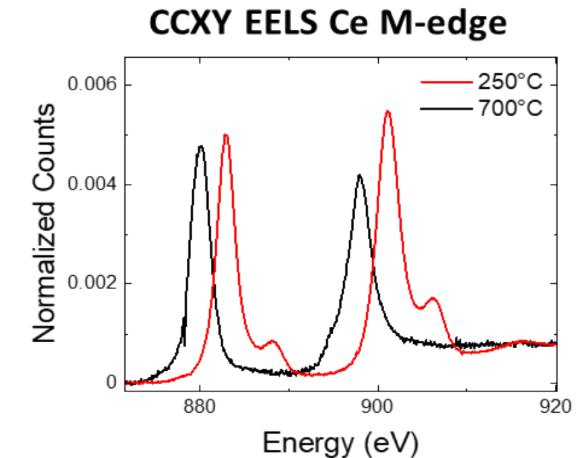
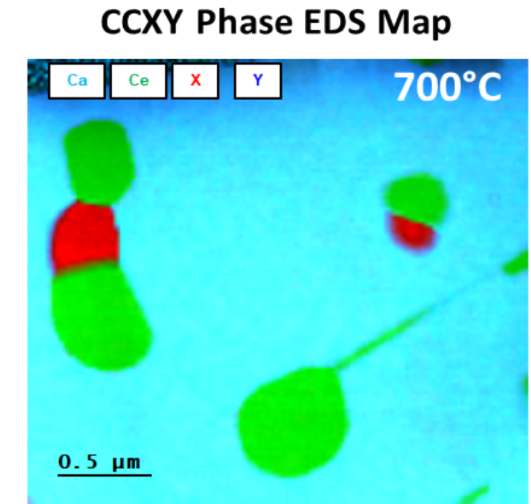
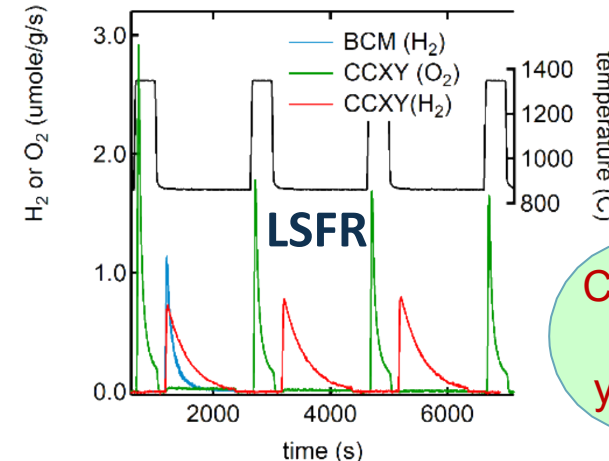
# ASU/Princeton Seedling Project: HydroGEN Node Support Provided by NREL (w/ SLAC) and SNL

- Predict new material family:  $\text{Ca}_{0.5}\text{Ce}_{0.5}\text{XO}_3$  with A-site redox activity.
  - <https://doi.org/10.1021/acs.chemmater.0c02912>
- NREL: Synthesized and characterized crystal structure and cation redox.
  - SLAC confirmed structure of predicted and enhanced stability material with cation Y substitution – “CCXY”
  - Confirmed dual-cation reduction mechanism during redox by XAS
- SNL: Characterized water splitting and A-site cation redox.
  - Confirmed CCXY splits water at low  $p_{\text{O}_2}$
  - Confirmed  $\text{Ce}^{(4+/3+)}$  redox in CCXY phase as predicted

**CCXY  $\text{H}_2$  prod capacity > SLMA >> BCM**



To=850C, 40%  $\text{H}_2\text{O}$  for 1200s, Tr=1350C for 330s



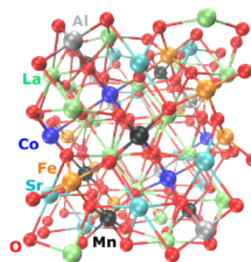
**CCXY exhibits high-yield water splitting with same relative yield as BCM (75%) @1333:1**



# UCSD Seedling Project: HydroGEN Node Support Provided by NREL and SNL

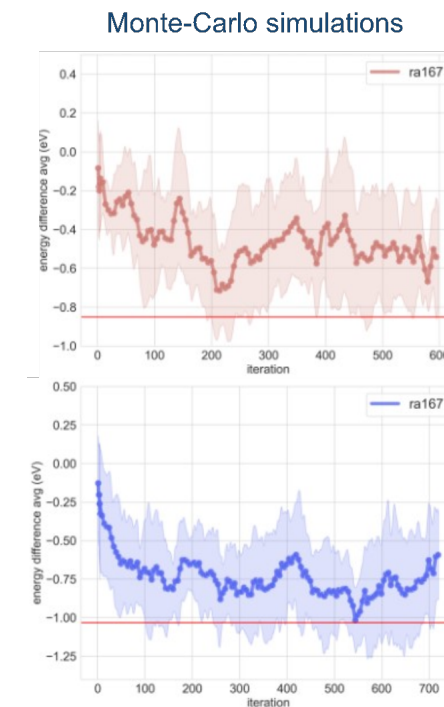
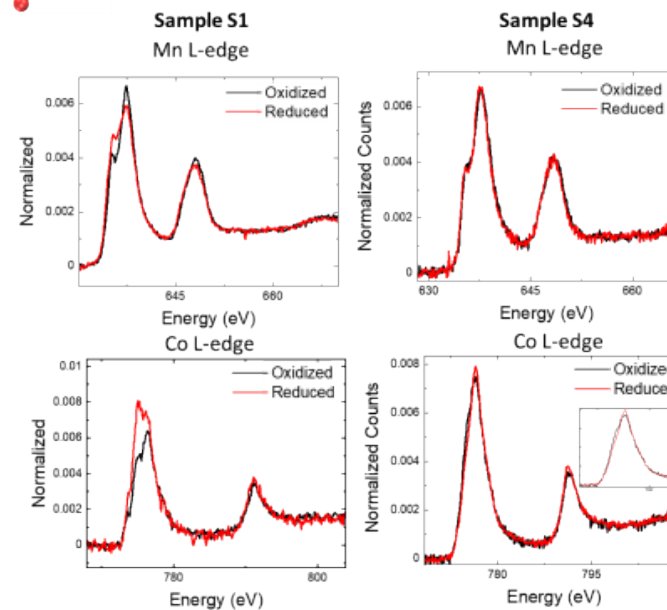
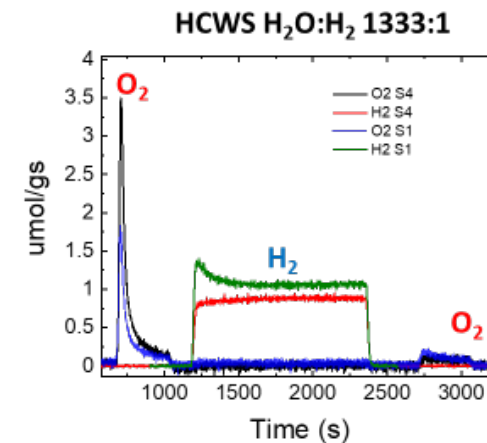
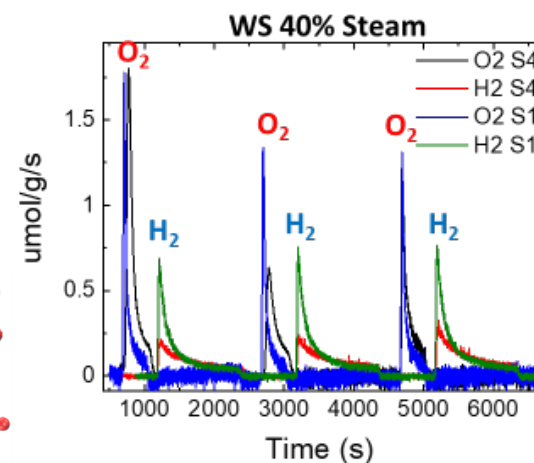
- Subtle compositional variation in  $\text{La}_{0.8}\text{Sr}_{0.2}(\text{Mn}_\alpha\text{Fe}_\alpha\text{Co}_{(0.16 \text{ or } 0.40)}\text{Al}_\alpha)\text{O}_3$  greatly affects redox behavior.

Sample compositions	
HEPO-S1	$(\text{La}_{0.8}\text{Sr}_{0.2})(\text{Mn}_{0.28}\text{Fe}_{0.28}\text{Co}_{0.16}\text{Al}_{0.28})\text{O}_3$
HEPO-S4	$(\text{La}_{0.8}\text{Sr}_{0.2})(\text{Mn}_{0.2}\text{Fe}_{0.2}\text{Co}_{0.4}\text{Al}_{0.2})\text{O}_3$



- Sampled configurational disorder via Monte-Carlo simulations for bulk and O vacancy defect structures.

- Short range ordering (SRO) is moderate in oxidized bulk
- Co is redox active in reduced oxide, preferential coordination of O vacancies





# Collaboration Effectiveness: Material Science Data and Expertise from Hydrogen

- Enable seedling projects to achieve project go/no-go milestones.
- Trained next generation STCH workforce.
  - Xin Qian and Rob Wexler former graduate students from STCH seedling projects
- Highly effective collaborations continue to produce joint publications.
  - R.B. Wexler, G.S. Gautam, R.T. Bell, S. Shulda, N.A. Strange, J.A. Trindell, J.D. Sugar, E. Nygren, S. Sainio, A.H. McDaniel, D. Ginley, E.A. Carter, E.B. Stechel, “Multiple and Nonlocal Cation Redox in Ca–Ce–Ti–Mn Oxide Perovskites for Solar Thermochemical Applications,” *Energy Environ. Sci.*, 2023. [10.1039/D3EE00234A](https://doi.org/10.1039/D3EE00234A)
  - D. Zhang, H.A. De Santiago, B. Xu, C. Liu, J.A. Trindell, Wei Li, J. Park, M.A. Rodriguez, E.N. Coker, J.D. Sugar, A.H. McDaniel, S. Lany, L. Ma, Y. Wang, G. Collins, H. Tian, W. Li, Y. Qi, X. Liu, and J. Luo, “Compositionally Complex Perovskite Oxides for Solar Thermochemical Water Splitting,” *Chemistry of Materials*, 2023. [10.1021/acs.chemmater.2c03054](https://doi.org/10.1021/acs.chemmater.2c03054)



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